

DESCRIPTION

FLUORINATED ALLYL ETHER POLYMER

FIELD OF THE INVENTION

- 5 The present invention relates to a fluorinated allyl ether polymer, in particular, a homo- or copolymer of 1,1,2-trifluoroallyl ether.

BACKGROUND ART

- 10 Hitherto, it is believed that hydrocarbon allyl compounds are hardly radically homopolymerized, and no homopolymer thereof having a high molecular weight can be obtained, and the reported molecular weight of the homopolymer is from about 300 to about 3,000 (see R. L. Shriner, L. Kelley ed. "Chemical Reviews" (USA), page 815, received by the National Diet Library (Japan) on December 15 13, 1962).

Thus, various improvements have been made on comonomers copolymerizable with the allyl compounds. However, no publications other than the above reference has reported the homo- and copolymerization of the allyl compounds.

20 DISCLOSURE OF THE INVENTION

An object of the present invention is to provide a polymer of a fluorinated allyl ether having a high molecular weight, which is prepared by radical homopolymerization.

- 25 This object is achieved by a fluorinated allyl ether polymer having a number average molecular weight of 1,000 to 1,000,000 and consisting of chains of at least one repeating unit of the formula:

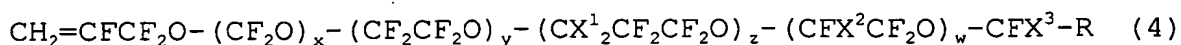
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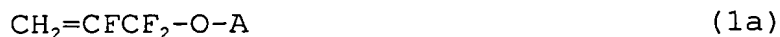


wherein A is an organic group having 1 to 100 carbon atoms, and
 5 a fluorinated allyl ether polymer having a number average
 molecular weight of 1,000 to 1,000,000 and represented by the
 formula:



wherein X^1 is a hydrogen atom, a fluorine atom or a chlorine atom,
 10 X^2 is a hydrogen atom, a chlorine atom, a methyl group or a
 trifluoromethyl group, X^3 is a hydrogen atom, a fluorine atom,
 a chlorine atom or a trifluoromethyl group, x , y , z and w are
 the same or different and a number of 0 to 20 provided that the
 sum of x , y , z and w is from 1 to 20, and R is $-\text{COOH}$, $-\text{COOR}^1$ in
 15 which R^1 is a hydrocarbon group having 1 to 20 carbon atoms, $-\text{CH}_2\text{OH}$,
 $-\text{CONH}_2$, $-\text{CF}=\text{CF}_2$, a hydrocarbon group having 1 to 20 carbon atoms
 or a perfluoroalkyl group having 1 to 20 carbon atoms.

The present invention is based on the finding that the
 specific allyl ether structure has good homopolymerizability and
 20 provides a polymer having a high molecular weight. That is, a
 fluorine-containing allyl ether compound of the formula:



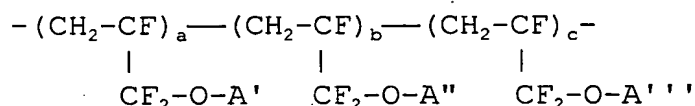
wherein A is an organic group having 1 to 100 carbon atoms has
 good homopolymerizability, in particular radical
 25 polymerizability, of the monomer of the formula (1a) due to the
 structure of the group: $\text{CH}_2=\text{CFCF}_2\text{-O-}$ in the formula (1a), and
 provides a high molecular weight polymer. This is particularly
 different from other allyl compounds such as $\text{CF}_2=\text{CFCF}_2\text{OR}$,
 $\text{CF}_2=\text{CFCF}_2\text{R}$, etc. or vinyl ether compounds such as $\text{CF}_2=\text{CFOR}$,

CH₂=CHOR, etc., which cannot be homopolymerized or which provide only low molecular weight products (oligomers), if they can be polymerized.

According to the present invention, insofar as the monomer
5 has the group of the formula: CH₂=CF CF₂-O-, it can be radically polymerized irrespective of the kind of the group A, and the polymer of the formula (1) can be obtained.

DETAILED DESCRIPTION OF THE INVENTION

The fluorinated allyl ether polymer of the present
10 invention consists of the chains of the repeating unit of the formula (1), and includes, for example, a copolymer of the formula:



15 In this formula, the order of the repeating units may be random or block.

The above polymer may be copolymerized with less than 20 %
by mole, preferably less than 10 % by mole of an ethylenically
unsaturated compound copolymerizable with the above fluorinated
20 allyl ether such as a fluorine-containing olefin.

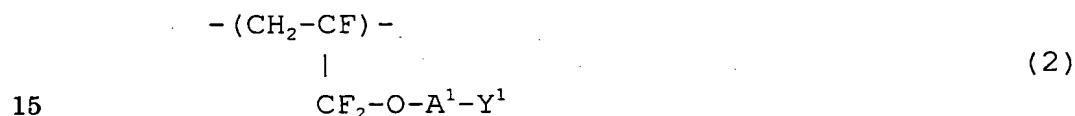
The group A may be selected from any organic groups. In
general, examples of the organic group include saturated or
unsaturated aliphatic hydrocarbon groups having 1 to 50 carbon
atoms, aromatic hydrocarbon groups having 4 to 30 carbon atoms,
25 etc. At least one of hydrogen atoms of these hydrocarbon groups
may be substituted with a fluorine atom.

Specific examples of the hydrocarbon groups include alkyl
or fluoroalkyl groups having 1 to 50 carbon atoms, alkenyl or
fluoroalkenyl groups having 2 to 50 carbon atoms, alkynyl or

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fluoroalkynyl groups having 2 to 50 carbon atoms, alkyl or fluoroalkyl groups having an ether bond and 1 to 60 carbon atoms, alkenyl or fluoroalkenyl groups having an ether bond and 2 to 60 carbon atoms, alkynyl or fluoroalkynyl groups having an ether bond and 2 to 60 carbon atoms, aryl or fluoroaryl groups having 4 to 30 carbon atoms, etc.

The hydrocarbon group may have a functional group. When the hydrocarbon group having the functional group is used, various functions are preferably imparted to the fluorinated allyl ether polymer obtained. A specific example of the repeating unit comprising the group A having the functional group is a repeating unit of the formula:



wherein A^1 is a divalent organic group having 1 to 60 carbon atoms, and Y^1 is $-\text{CH}_2\text{OH}$, $-\text{COOH}$, $-\text{COOR}^1$ in which R^1 is a hydrocarbon group having 1 to 20 carbon atoms, $-\text{CON} \begin{smallmatrix} \text{R}^2 \\ < \\ \text{R}^3 \end{smallmatrix}$ in which R^2 and R^3 are the same or different and a hydrogen atom or a hydrocarbon group having 1 to 20 carbon atoms, $-\text{O}-\text{CF}=\text{CF}_2$, or $-\text{OCO}-\text{CZ}^3=\text{CZ}^1\text{Z}^2$ in which Z^1 and Z^2 are the same or different and a hydrogen atom or a fluorine atom, and Z^3 is a hydrogen atom, a fluorine atom, a chlorine atom or a trifluoromethyl group.

Besides the group $-\text{A}^1-\text{Y}^1$, examples of the group A having the functional group include an epoxy group, a glycidyl group, a cyano group ($-\text{CN}$), a sulfonic acid group ($-\text{SO}_3\text{H}$), $-\text{SO}_3\text{R}'$ in which R' is a monovalent organic group, etc.

The group A^1 in the formula (2) may be selected from divalent organic groups having 1 to 60 carbon atoms, preferably, divalent

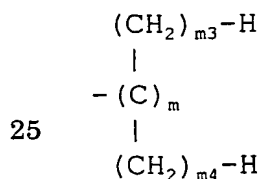
fluoroalkylene groups, divalent fluoroalkylene group having an ether bond, etc. These groups can impart various functions such as heat resistance, stain-proofing, non-tackiness, optical properties (e.g. low refractive index), chemical resistance, 5 electrical insulation, etc. to the polymer.

Specific examples of the divalent fluoroalkylene group include $-(CF_2)_m-(CH_2)_n-$, $-[CF_2CF(CF_3)]_m-(CH_2)_n-$, $-(CF_2CH_2)_m-(CH_2)_n-$, and $-[CF_2C(CF_3)_2]_m-(CH_2)_n-$ wherein m is a number of 1 to 20, and n is a number of 0 to 10.

10 Specific examples of the divalent fluoroalkylene group having the ether bond include $-(CF_2CF_2O)_m-CF_2-$, $-(CF_2CF_2CF_2O)_m-CF_2CF_2-$, $-[CF(CF_3)CF_2O]_m-CF(CF_3)-$, and $-(CF_2O)_m-(CF_2)_k-$ wherein m and k are the same or different and a number of 1 to 20.

Besides the above fluoroalkylene groups and the like, the 15 group A¹ in the formula (2) may be one of the following hydrocarbon groups having 1 to 20 carbons:

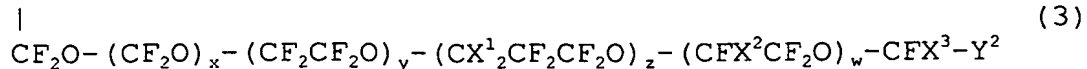
$-(CH_2)_m-$ wherein m is a number of 1 to 20, a cyclohexylene group, $-(Ph)_a-$ wherein Ph is a phenylene group and a is an integer of 1 to 3, $-(CH_2)_{m1}-(Ph)_a-(CH_2)_{m2}-$ in which Ph and a are the same as 20 defined above, m₁ is a number of 1 to 5 and m₂ is a number of 0 to 5,



wherein m is a number of 1 to 20, m₃ is a number of 0 to 10 and m₄ is a number of 1 to 10, and the like.

One particularly preferable example of the repeating unit 30 of the formula (1) is a repeating unit of the formula:

-CH₂-CF-



wherein X¹ is a hydrogen atom, a fluorine atom or a chlorine atom,

- 5 X² is a hydrogen atom, a chlorine atom, a methyl group or a trifluoromethyl group, X³ is a hydrogen atom, a fluorine atom, a chlorine atom or a trifluoromethyl group, x, y, z and w are the same or different and a number of 0 to 20 provided that the sum of x, y, z and w is from 1 to 20, and Y² is -COOH, -COOR⁴ in
10 which R⁴ is a hydrocarbon group having 1 to 20 carbon atoms, -CH₂OH, -CON^{R⁵}_{R⁶} in which R⁵ and R⁶ are the same or different and a hydrogen atom or a hydrocarbon group having 1 to 20 carbon atoms, -O-
15 CF=CF₂, or -OCO-CZ⁶=CZ⁴Z⁵ in which Z⁴ and Z⁵ are the same or different and a hydrogen atom or a fluorine atom, and Z⁶ is a hydrogen atom, a fluorine atom, a chlorine atom or a trifluoromethyl group.

Preferable examples of the fluorinated allyl ether (3)

- 20 include CH₂=CF CF₂OCF(CF₃)CF₂OCF(CF₃)COOCH₃,
CH₂=CF CF₂OCF(CF₃)CF₂OCF(CF₃)CF₂OCF(CF₃)COOCH₃,
CH₂=CF CF₂OCF(CF₃)CF₂OCF(CF₃)COOH,
CH₂=CF CF₂OCF(CF₃)CF₂OCF(CF₃)CF₂OCF(CF₃)COOH,
CH₂=CF CF₂OCF(CF₃)CF₂OCF(CF₃)CH₂OH,
25 CH₂=CF CF₂OCF(CF₃)CF₂OCF(CF₃)CF₂OCF(CF₃)CH₂OH, etc.

The preparation method of the fluorinated allyl ether (3) of the present invention will be explained. To simplify the formulas, "-(CF₂O)_x-(CF₂CF₂O)_y-(CX¹₂CF₂CF₂O)_z-(CFX²CF₂O)_w-CFX³-" will be represented by "Rf" in the following description.

- 30 The fluorinated allyl ether (3) of the present invention can be easily synthesized by defluoriodination of a compound

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having a terminal iodine atom of the formula:



The fluoroiodination is preferably carried out in a solvent (e.g. dimethylformamide, dimethylsulfoxide, methanol, acetone, methyl
5 ethyl ketone, ethyl acetate, etc.) in the presence of a catalyst (e.g. zinc, copper, etc.) at a temperature of -20 to 200°C, preferably 50 to 150°C.

The polymer consisting of the chains of the repeating unit of the formula (1) of the present invention may be prepared as
10 follows:

The homopolymerization of the fluorinated allyl ether (1) of the present invention is carried out by a conventional radical polymerization method under conventional radical polymerization conditions. Preferably, the radical polymerization using a
15 radical initiating source (e.g. a radical polymerization initiator, light, heat, etc.) is used, and the polymerization mode may be bulk polymerization, solution polymerization, emulsion polymerization, suspension polymerization, etc.

The conditions employed in the radical polymerization are
20 not limited and may include a temperature of 0 to 100°C and a pressure of atmospheric pressure, a reduced pressure down to about 760 mmHg or an elevated pressure up to about 100 kg/cm².

Alternatively, a functional group can be introduced in the polymer obtained by reacting the polymer with a compound having
25 a functional group.

For example, when a polymer, which is obtained by polymerizing the fluorinated allyl ether (4) having the -CH₂OH group, is reacted with a compound of the formula: $\text{CZ}^1\text{Z}^2=\text{CZ}^3\text{COF}$

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wherein Z^1 and Z^2 are the same as defined above, a group of the formula: $-\text{CH}_2\text{OCO}-\text{CZ}^3=\text{CZ}^1\text{Z}^2$ can be introduced in the polymer.

Alternatively, when a polymer obtained by polymerizing the fluorinated allyl ether (4) having the $-\text{OCF}(\text{CF}_3)\text{COOH}$ group is
 5 treated with sodium hydroxide, etc. to change this group to an alkali salt and then heated to decarboxylate the polymer, the $-\text{OCF}=\text{CF}_2$ group can be introduced in the polymer.

The fluorinated allyl ether of the present invention can provide a high molecular weight product through the radical
 10 homopolymerization, and when the fluorinated allyl ether having the functional group is used, the polymer having the functional group in the molecule can be easily obtained. Such polymers have various properties depending on the functional groups contained therein, and they may be used as solvent-resistant polymers,
 15 water-soluble polymers, ion-exchange resins, reactive polymers, etc.

The present invention will be explained in detail by the following examples.

Example 1

20 $\text{CH}_2=\text{CFCF}_2\text{OCF}(\text{CF}_3)\text{CF}_2\text{OCF}(\text{CF}_3)\text{COOCH}_3$ (5 g) and a 8 wt.% solution of $[\text{H}(\text{CF}_2\text{CF}_2)_3\text{COO-}]_2$ (hereinafter referred to as "DHP") in trichlorotrifluoroethane (0.5 g) were charged in a glass vessel. When the internal atmosphere in the vessel was replaced with nitrogen and the mixture was stirred at room temperature for 24
 25 hour, the viscosity of the mixture increased. Low boiling materials were distilled off from the reaction mixture under reduced pressure to obtain a colorless transparent polymer (4.67 g).

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When a molecular weight was calculated on the assumption that the initiator efficiency was 1 (one), no chain transfer reaction took place, and the termination was only rebonding termination, it was 71,652. Hereinafter, the molecular weight
5 calculated as above will be referred to as a "calculated molecular weight".

The polymer obtained was soluble in tetrahydrofuran (THF). The polymer was dissolved in THF, and its molecular weight was measured with GPC using polystyrene as a standard. A number
10 average molecular weight was about 68,000. This number average molecular weight measured with GPC was in good agreement with the calculated molecular weight. Thus, in the following Examples, only calculated molecular weights are reported.

The polymer had Tg of -2°C and a refractive index of 1.3132.
15 Herein, Tg was measured with DSC (differential scanning calorimeter), and a refractive index was measured with an Abbe refractometer.

Example 2

$\text{CH}_2=\text{CFCF}_2\text{OCF}(\text{CF}_3)\text{CF}_2\text{OCF}(\text{CF}_3)\text{CF}_2\text{OCF}(\text{CF}_3)\text{CH}_2\text{OH}$ was polymerized
20 in the same manner as in Example 1, and a colorless transparent polymer (2.10 g) was obtained. It had a calculated molecular weight of 32,865, Tg of -4°C and a refractive index of 1.3416.

Example 3

$\text{CH}_2=\text{CFCF}_2\text{OCF}(\text{CF}_3)\text{CF}_2\text{OCF}(\text{CF}_3)\text{COOH}$ was polymerized in the same
25 manner as in Example 1, and a white polymer (4.77 g) was obtained. It had a calculated molecular weight of 74,651 and Tg of 13°C. This polymer was soluble in water, and pH of the aqueous solution (concentration: 1 wt. %) was about 2. The aqueous solution was

foamable.

Example 4

Polymerization was carried out in the same manner as in Example 1 except that $\text{CH}_2=\text{CFCF}_2\text{OCF}(\text{CF}_3)\text{CF}_2\text{OCF}(\text{CF}_3)\text{CH}_2\text{OH}$ (5.0 g) and DHP (0.51 g) were used, and a colorless transparent polymer (4.68 g) was obtained. This polymer was a hard solid at room temperature, and it had a calculated molecular weight of 80,730.

Example 5

Polymerization was carried out in the same manner as in Example 1 except that $\text{CH}_2=\text{CFCF}_2\text{OCF}(\text{CF}_3)\text{CF}_2\text{OCF}(\text{CF}_3)\text{CF}_2\text{OCF}(\text{CF}_3)\text{COOCH}_3$ (5.01 g) and DHP (0.53 g) were used, and a colorless transparent polymer (4.54 g) was obtained. This polymer was soft at room temperature, and it had a calculated molecular weight of 73,880.

Example 6

$\text{CH}_2=\text{CFCF}_2\text{OCF}(\text{CF}_3)\text{CF}_2\text{OCF}(\text{CF}_3)\text{CH}_2\text{OH}$ (Monomer A) (14. g), $\text{CH}_2=\text{CFCF}_2\text{OCF}(\text{CF}_3)\text{CF}_2\text{OCF}(\text{CF}_3)\text{COOCH}_3$ (Monomer B) (6 g) and DHP (8.29 g; 8 wt. % solution in perfluorohexane) were mixed with HCFC-225 (a mixture of 1,1-dichloro-2,2,3,3,3-pentafluoropropane and 1,3-dichloro-1,2,2,3,3-pentafluoropropane) (5 g), and the mixture was stirred at 30°C for 24 hours.

The resulting solution was poured in petroleum benzine to precipitate the polymer formed.

The polymer precipitated was recovered, washed with petroleum benzine, and dried under reduced pressure to obtain a rubbery polymer (18 g).

According to NMR analysis, the polymer obtained was a copolymer containing Monomer A and Monomer B in a molar ratio of 81:19.

Example 7

The fluorine-containing copolymer obtained in Example 6 (13 g) was dissolved in HCFC-225 (30 g), and the pyridine (0.61 g) was added to the solution, followed by cooling to 0 to 5°C. Then, 5 $\text{CH}_2=\text{CFCOF}$ (3.0 g) was gradually dropwise added to the solution while cooling with ice and stirring, and then the solution was stirred for additional 4 hours.

The polymer was precipitated washed and dried in the same manner as in Example 6 to obtain a rubbery polymer (10 g).

10 In the NMR chart of the obtained polymer, the absorption assigned to the OH groups disappeared, while the absorption assigned to $>\text{C}=\text{C}<$ appeared.

According to NMR analysis, in the polymer obtained, the molar ratio of the monomer having $-\text{COOCH}_3$ to the monomer having 15 $-\text{OCO}-\text{CF}=\text{CH}_2$ was 80:20.

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